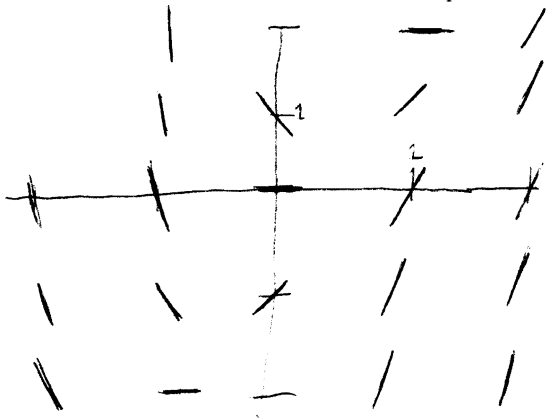


Answer the following questions showing all work. 5 points each.

1. Let $\frac{dy}{dx} = 2x - y$. Sketch a slope field for this differential equation. Is $y = 2x$ a solution to the differential equation? Why or why not?



No, because if $y = 2x$, then $\frac{dy}{dx} = 2$; but $2x - y = 2x - (2x) = 0$.

So $\frac{dy}{dx} \neq 2x - y$

2. Is $y = e^{e^x}$ a solution to $\frac{dy}{dx} = \ln(y^y)$? [HINT: It may help to simplify first. Think about exponent rules.]

Suppose $y = e^{e^x}$. Then by chain rule, $\frac{dy}{dx} = e^x \cdot e^{e^x}$. Also, $\ln(y^y) = y \ln y = e^{e^x} \cdot \ln(e^{e^x}) = e^{e^x} \cdot e^x$. Thus $\frac{dy}{dx} = \ln(y^y)$, so it is a soln

Alternatively, $\frac{dy}{dx} = \ln(y^y) = y \ln y$
 $\frac{dy}{y \ln y} = dx$

$\int \frac{1}{y \ln y} dy = \int dx$
 $\ln |\ln y| = x + C$
 w-sub w/ $w = \ln y$ & $dw = \frac{1}{y} dy$

$|\ln y| = e^{x+C} \Rightarrow B = e^C$
 $\ln y = B e^x$
 $y = e^{(B e^x)}$ ← This is all solutions to this diff eq; we have the one where $B = 1$

3. Let $\frac{dq}{dt} = q^2 t$ and assume $q(0) = 1$. Using Euler's method with a Δt step of .5, estimate $q(1)$.

@ $t = 0$ $q = 1$ & $\frac{dq}{dt} = q^2 t = 1^2 \cdot 0 = 0 = m$ So our tangent line is $q \approx 0(t - 0) + 1$, which gives us $q(0.5) \approx 0(0.5 - 0) + 1 = 1$

@ $t = .5$ $q \approx 1$, so $\frac{dq}{dt} = (1)^2 (.5) = .5 = m$. So our tangent line is $q \approx .5(t - .5) + 1$, which gives $q(1) \approx .5(1 - .5) + 1 = 1.25$

So $q(1) \approx 1.25$

4. The most interesting man in the world wants a refreshing beer after a long day of wrestling tigers. Unfortunately, his liquid nitrogen chamber has no beer in it. So he puts a room temperature ($70^\circ F$) Dos Equis in. One second later, the temperature has already dropped to $69^\circ F$. Set up and solve a differential equation to determine how long the beer needs to be before it is cold enough to drink (the most interesting man in the world prefers his beers to be $34^\circ F$). You may assume Newton's Law of Cooling¹ and his liquid nitrogen chamber is kept at a chilly $-350^\circ F$.

$$\frac{dT}{dt} = k(T_{beer} - T_{air})$$

$$\frac{dT}{dt} = k(T + 350)$$

$$\int \frac{dT}{T+350} = \int k dt$$

$$\ln|T+350| = kt + C$$

$$|T+350| = e^{kt+C}$$

$$T+350 = Be^{kt}$$

$$T = Be^{kt} - 350$$

@ $t=0$ $70 = Be^{0} - 350$
 $420 = B$

@ $t=1$ $69 = Be^{k} - 350$
 $419 = 420e^k$
 $k = \ln\left(\frac{419}{420}\right) \approx -0.00238$

$34 = 420e^{t(-0.00238)} - 350$
 $384 = 420e^{-0.00238t}$
 $-0.00238t = \ln\left(\frac{384}{420}\right) \approx -0.0896$

$t = \frac{-0.0896}{-0.00238} \approx \underline{\underline{37.65 \text{ seconds!}}}$

5. BONUS: Willy Wonka's 1000 gallon chocolate swimming pool starts with 100 gallons of water in it with a chocolicity (chocolate concentration) of 20 grams of chocolate per gallon. Fresh water is flowing into the tank at a rate of 10 gallons per minute, while 1 gallon of the mixed chocolate water leaks out every minute. What will the chocolicity of the water in the tank be when the tank is full?

Let $M(t)$ = Mass of chocolate @ time t
 Let $V(t)$ = Volume of H_2O @ time t
 Let $C(t)$ = Chocolicity @ time t
 So $C(t)V(t) = M(t)$ $[CV=M]$

$$V(t) = 100 + (10 - 1)t$$

$$V(t) = 100 + 9t$$

$$\frac{dM}{dt} = \frac{-M}{100+9t}$$

$$\int \frac{-dM}{M} = \int \frac{-dt}{100+9t}$$

$$-\ln|M| = \frac{1}{9} \ln|100+9t| + C$$

$$-\ln M = \frac{1}{9} \ln(100+9t) + C$$

$$e^{-\ln M} = e^{\frac{1}{9} \ln(100+9t) + C}$$

Let $B = e^C$

$$M^{-1} = B(100+9t)^{1/9} \rightarrow D = \frac{1}{B}$$

$$M = D(100+9t)^{-1/9}$$

@ $t=0$, we have $M = 20 \cdot 100 = 2000$
 $\rightarrow 2000 = D(100+0)^{-1/9} \rightarrow$
 $D = 2000(100)^{1/9} \approx 1199$

Tank full when $V = 1000 = 100 + 9t$
 tank full when $t = 100$

$$M = \frac{1199}{(1000)^{1/9}} \approx 557.67$$

$$C = \frac{M}{V} = \frac{557.67}{1000} \approx 0.557 \frac{g}{gal}$$

¹The rate of change of the temperature of a body is directly proportional to the difference between the temperature of the body and the temperature of the surrounding air.

* b, c M & $100+9t$
 both always positive